

ATOMIC ENERGY EDUCATION SOCIETY

Distant Learning Programme

Class XI

Subject: Physics

Hand out study Material

Chapter: Unit and Measurement (Module 1/4)

Contents:

- Physical Quantities and their types
- Measurement and unit system
- Types of unit system
- Advantage of SI unit system
- Definition of fundamental units
- Measurement of length (Parallax method)
- Some special length units
- Measurement of mass and time
- Physical Quantities and their types

•Physical Quantities

- Those quantities which can be measured and, are necessary to describe any physical phenomenon.
- There are two types of physical quantities:
- Fundamental physical quantities: Those quantities which are independent and do not depend on other quantities are called as fundamental quantities.
- Mass, Length and Time are called Fundamental physical quantities
- Derived Physical quantities: Those physical quantities which can be expressed in terms of fundamental physical quantities.
- Ex. Area, volume, force, work, pressure etc.

** For detailed list of derived physical quantities refer to appendix A6.1, A6.2 and A6.3 of NCERT text book part 1

SI unit system

Internationally accepted system of units is **Système International d' Unites (French for International system of Units) or SI**. It was developed and recommended by **General Conference on Weights and Measures in 1971**.

- SI lists 7 base units as in the table below. Along with it, there are two units - radian or rad (unit for plane angle) and steroidal or sr (unit for solidangle). They both are dimensionless.

Base Quantity	SI Unit	Symbol
Length	Meter	m
Mass	Kilogram	kg
Time	Second	s
Electric current	Ampere	A or I
Temperature	Kelvin	K
Luminous Intensity	Candela	cd
Amont of substance	mol	mol

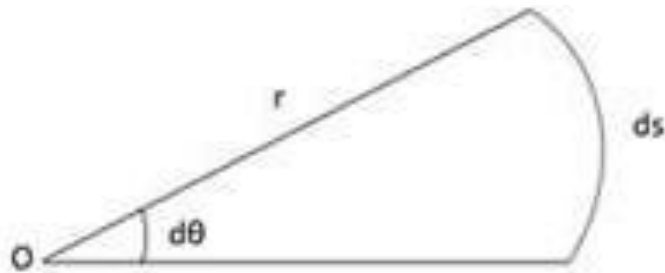
Advantages of SI unit system

- SI is a coherent system of units i.e system based on a certain set of fundamental units, from which all derived units are obtained by multiplication or division without introducing numerical factors.
- SI is a rational system of units as it assigns only one unit to be a particular physical quantity. For example, joule is the unit for all types of energy. This is not so in other systems of units.
- SI is an absolute system of units. There are no Gravitational units on the system. This use factor ‘g’ is thus eliminated.
- SI is a metric system i.e the multiples and sub multiples of units are expressed as power of 10.

Definition of fundamental units

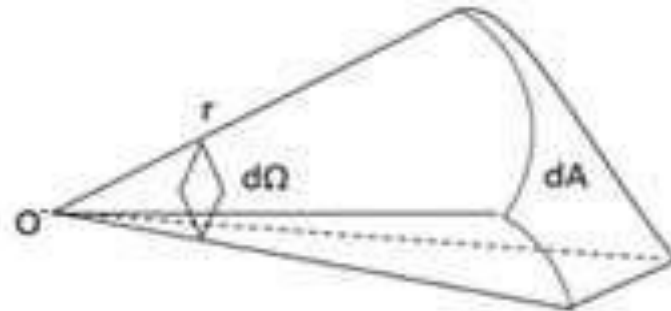
- **1 meter :** The meter is the length of the path travelled by light in vacuum during a time interval of $1/299792\ 458$ of a second.
- **1 kilogram:** The kilogram is the unit of mass; it is equal to the mass of the international prototype of the kilogram. The international prototype of the kilogram, an artifact made of Platinum-iridium is kept at the BIPM under the conditions specified by the 1st CGPM in 1889 (CR, 34-38) when it sanctioned the prototype and declared:
- **1 second:** The second is the duration of $9\ 192\ 631\ 770$ periods of the radiation corresponding to the transition between the two hyperfine levels of the ground state of the cesium 133 atom.
- **1 Ampere:** The ampere is that constant current which, if maintained in two straight parallel conductors of infinite length, of negligible circular cross section, and placed 1 meter apart in vacuum, would produce between these conductors a force equal to 2×10^{-7} Newton per meter of length.
- **1 Kelvin:** The Kelvin, unit of thermodynamic temperature, is the fraction $1/273.16$ of the thermodynamic temperature of the triple point of water.
- **1 Candela:** The candela is the luminous intensity, in a given direction, of a source that emits monochromatic radiation of frequency 540×10^{12} hertz and that has a radiant intensity in that direction of $1/683$ watt per steradian.
- **1 mole:** The mole is the amount of substance as there are atoms in 0.012 kilogram of carbon 12; its symbol is “mol.”

1 Radian & 1 steradian :



$$d\theta = ds/r \text{ radian}$$

Plane angle. Unit - Radian



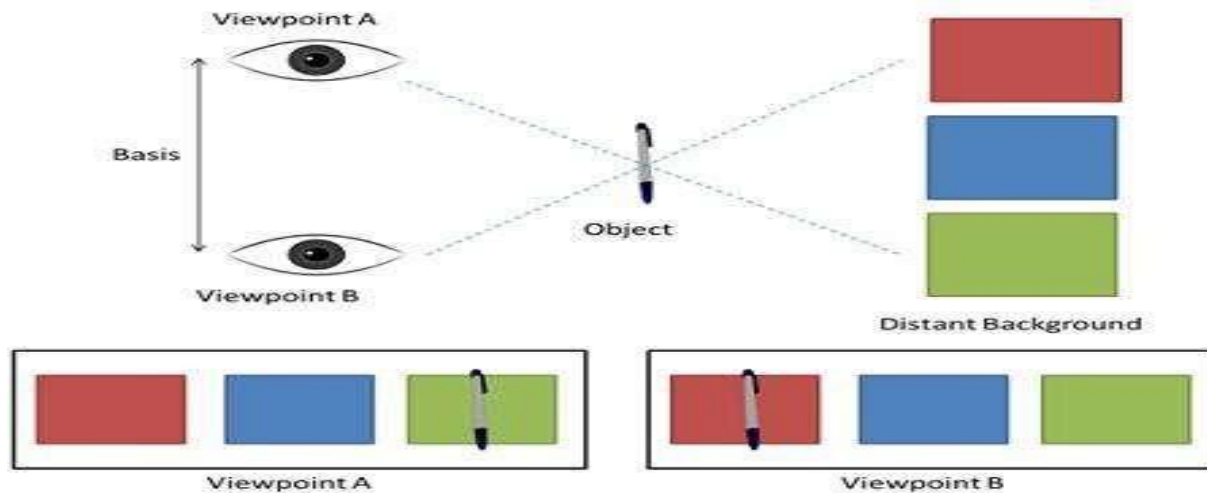
$$d\Omega = dA/r^2 \text{ steradian}$$

Solid angle. Unit - Steradian

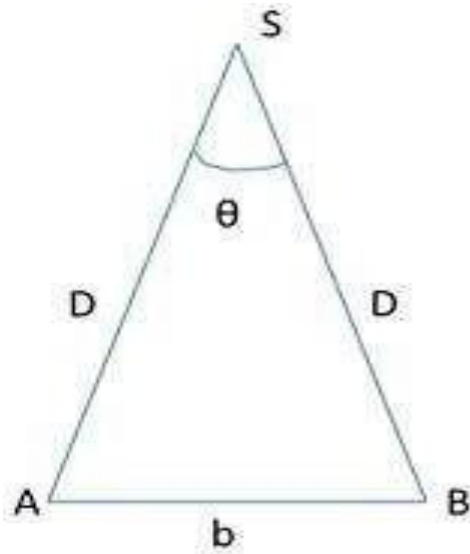
Measurement of length

Measuring large Distances – Parallax Method

- Parallax is a displacement or difference in the apparent position of an object viewed along two different lines of sight, and is measured by the angle or semi-angle of inclination between those two lines. Distance between the two viewpoints is called Basis.



Parallax. From viewpoint A the pen appears over green box while from viewpoint B the pen appears over red box.



S – position of the planet
 D – Distance from the two viewpoints or observatories
 θ – parallax or parallactic angle

For far away planet, $b/D \ll 1$
 Hence, AB is taken as an arc of length b and D is radius with S as center.
 So, $b = D\theta$ or $D = b/\theta$

Parallax method to determine distance of a planet

Measuring distance of a planet using parallax method:

Similarly, $\alpha = d/D$

Where α = angular size of the planet (angle subtended by d at earth) and d is the diameter of the planet. α is angle between the direction of the telescope when two diametrically opposite points of the planet are viewed.

Measuring very small distances

To measure distances as low as size of a molecule, **electron microscopes are used. These contain electrons beams controlled by electric and magnetic fields.**

- Electron microscopes have a resolution of 0.6 \AA or Angstroms.
- Electron microscopes are able to resolve atoms and molecules while using tunneling microscopy, it is possible to estimate size of molecule.

Some special length units

Unit name	Unit Symbol	Value in meters
fermi	f	10-15 m
angstrom	Å	10-10m
astronomical unit(average distance of sun from earth)	AU	1.496 X 10 ¹¹ m
light year(distance travelled by light in 1 year with velocity 3×10⁸m/s)	ly	9.46 X 10 ¹⁶ m
parsec(distance at which average radius of earth's orbits subtends an angle of 1 arc second)	pc	3.08 x 10 ¹⁶ m

Measurement of Mass

- Mass is usually measured in terms of kg but for atoms and molecules, **unified atomic mass unit (u) is used**.
1 u = 1/12 of the mass of an atom of carbon-12 isotope including mass of electrons (1.66×10^{-27} kg)
- Apart from using **balances for normal weights, mass of planets is measured using gravitational methods and mass of atomic particles are measured using mass spectrograph** (radius of trajectory is proportional to mass of charged particle moving in uniform electric and magnetic field).

Measurement of Time

- Time is measured using a clock. As a standard, **atomic standard of time is now used, which is measured by Cesium or Atomic clock**.
 - In Cesium clock, a second is equal to 9,192,631,770 vibrations of radiation from the transition between two hyperfine levels of cesium-133 atom.
 - Cesium clock works on the vibration of cesium atom which is similar to vibrations of balance wheel in a regular wristwatch and quartz crystal in a quartz wristwatch.
 - National standard time and frequency is maintained by 4 atomic clocks. Indian standard time is maintained by a Cesium clock at National Physical Laboratory (NPL), New Delhi. Cesium clocks are very accurate and the uncertainty is very low 1 part in 10¹³ which means not more than 3 μ s are lost or gained in a year.
- ** Refer to table no. 2.4 & 2.5 of NCERT part 1 chapter 2 for various ranges of mass and time .**

**REFERNECES:
NCERT XI CLASS WIKIPEDIA
CONCEPT OF PHYSICS BY H C VERMA**

By: Govind Sharma
PGT (Physics) AECS 4, Rawatbhata